



*Annual*  
**WATER  
QUALITY  
REPORT**

*Reporting Year 2013*



*Presented By* \_\_\_\_\_  
**City of Cadillac**

PWS ID#: 0001030

## There When You Need Us

We are once again proud to present our annual water quality report covering all testing performed between January 1 and December 31, 2013. Over the years, we have dedicated ourselves to producing drinking water that meets all state and federal standards. We continually strive to adopt new methods for delivering the best-quality drinking water to you. As new challenges to drinking water safety emerge, we remain vigilant in meeting the goals of source water protection, water conservation, and community education while continuing to serve the needs of all our water users.

Please remember that we are always available to assist you should you ever have any questions or concerns about your water.



## Community Participation

We want to inform our customers about their water utility. Copies of our operation budget and capital improvement plan are available at the municipal complex and public library. If you would like to tour a facility or learn more about our operations, please call our office to make arrangements. City council meetings are another good public forum for community participation; feel free to attend one of our regularly scheduled city council meetings on the first and third Mondays of each month beginning at 7 p.m. at the Municipal Complex, 200 Lake Street, Cadillac, Michigan.

## Substances That Could Be in Water

To ensure that tap water is safe to drink, the U.S. EPA prescribes regulations limiting the amount of certain contaminants in water provided by public water systems. U.S. Food and Drug Administration regulations establish limits for contaminants in bottled water, which must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of these contaminants does not necessarily indicate that the water poses a health risk.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals, in some cases, radioactive material, and substances resulting from the presence of animals or from human activity. Substances that may be present in source water include:

**Microbial Contaminants**, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, or wildlife;

**Inorganic Contaminants**, such as salts and metals, which can be naturally occurring or may result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming;

**Pesticides and Herbicides**, which may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses;

**Organic Chemical Contaminants**, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and may also come from gas stations, urban stormwater runoff, and septic systems;

**Radioactive Contaminants**, which can be naturally occurring or may be the result of oil and gas production and mining activities.

For more information about contaminants and potential health effects, call the U.S. EPA's Safe Drinking Water Hotline at (800) 426-4791.

## Important Health Information

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants may be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. The U.S. EPA/CDC (Centers for Disease Control and Prevention) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791 or <http://water.epa.gov/drink/hotline>.

## Lead in Home Plumbing

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high-quality drinking water, but we cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at [www.epa.gov/safewater/lead](http://www.epa.gov/safewater/lead).



## QUESTIONS?

For more information about this report, or for any questions relating to your drinking water, please call the Cadillac Utilities Department at (231) 775-0181.

## Where Does My Water Come From?

Cadillac's water comes from eight water wells owned by the City. Our wells draw groundwater from aquifers 300 and 400 feet below ground. The City's older well field and 1-million-gallon water tower were constructed in 1960, ending our reliance on surface water from Lake Cadillac. The most current well field, consisting of three wells, was completed and put online in August of 2012.

Because well water contains varying amounts of iron, a polyphosphate solution is added at each well to try to reduce iron oxide formation in the water system. Chlorine is added to our system to disinfect the water supply.

## Source Water Assessment

The 1996 amendments to the Safe Drinking Water Act require that source water assessments be completed for all public water supplies in the United States. In our state, the Michigan Department of Natural Resources and Environment (MDNRE) developed a program to (1) identify areas that supply public drinking water, (2) assess the susceptibility of that supply to actual and potential contamination, and (3) inform the public of the assessment results. Cadillac's assessment was reevaluated in 2005 based on Cadillac's approved Wellhead Protection Program. MDNRE's revised assessment lists the wells with a high to very high susceptibility (based on geology), well construction, well-water chemistry, source isolation, and potential sources of contamination. Copies of the complete source water assessment are available at Cadillac's Municipal Complex and local DNRE office. To learn more about Cadillac's Wellhead Protection Program, please visit our Web site at [www.cadillac-mi.net](http://www.cadillac-mi.net).



## How Long Can I Store Drinking Water?

The disinfectant in drinking water will eventually dissipate even in a closed container. If that container housed bacteria before it was filled with the tap water, the bacteria may continue to grow once the disinfectant has dissipated. Some experts believe that water could be stored up to six months before needing to be replaced. Refrigeration will help slow the bacterial growth.

## What's a Cross-Connection?

Cross-connections that contaminate drinking water distribution lines are a major concern. A cross-connection is formed at any point where a drinking water line connects to equipment (boilers), systems containing chemicals (air conditioning systems, fire sprinkler systems, irrigation systems), or water sources of questionable quality. Cross-connection contamination can occur when the pressure in the equipment or system is greater than the pressure inside the drinking water line (backpressure). Contamination can also occur when the pressure in the drinking water line drops due to fairly routine occurrences (main breaks, heavy water demand), causing contaminants to be sucked out from the equipment and into the drinking water line (backsiphonage).

Outside water taps and garden hoses tend to be the most common sources of cross-connection contamination at home. The garden hose creates a hazard when submerged in a swimming pool or when attached to a chemical sprayer for weed killing. Garden hoses that are left lying on the ground may be contaminated by fertilizers, cesspools, or garden chemicals. Improperly installed valves in your toilet could also be a source of cross-connection contamination.

Community water supplies are continuously jeopardized by cross-connections unless appropriate valves, known as backflow prevention devices, are installed and maintained. We have surveyed all industrial, commercial, and institutional facilities in the service area to make sure that all potential cross-connections are identified and eliminated or protected by a backflow preventer. We also inspect and test each backflow preventer to make sure that it is providing maximum protection.

For more information, review the Cross-Connection Control Manual from the U.S. EPA's Web site at <http://water.epa.gov/infrastructure/drinkingwater/pws/crossconnectioncontrol/index.cfm>. You can also call the Safe Drinking Water Hotline at (800) 426-4791.



## Water Main Flushing

Distribution mains (pipes) convey water to homes, businesses, and hydrants in your neighborhood. The water entering distribution mains is of very high quality; however, water quality can deteriorate in areas of the distribution mains over time. Water main flushing is the process of cleaning the interior of water distribution mains by sending a rapid flow of water through the mains.

Flushing maintains water quality in several ways. For example, flushing removes sediments like iron and manganese. Although iron and manganese do not pose health concerns themselves, they can affect the taste, clarity, and color of the water. In addition, sediments can shield microorganisms from the disinfecting power of chlorine, contributing to the growth of microorganisms within distribution mains. Flushing helps remove stale water and ensures the presence of fresh water with sufficient dissolved oxygen and disinfectant levels, and an acceptable taste and smell.

During flushing operations in your neighborhood, some short-term deterioration of water quality, though uncommon, is possible. You should avoid tap water for household uses at that time. If you do use the tap, allow your cold water to run for a few minutes at full velocity before use and avoid using hot water, to prevent sediment accumulation in your hot water tank.

Please contact us if you have any questions or if you would like more information on our water main flushing schedule.

## Sampling Results

During the past year, we have taken hundreds of water samples in order to determine the presence of any radioactive, biological, inorganic, volatile organic, or synthetic organic contaminants. The tables below show only those contaminants that were detected in the water.

The state requires us to monitor for certain substances less often than once per year because the concentrations of these substances do not change frequently. In these cases, the most recent sample data are included, along with the year in which the sample was taken.

### REGULATED SUBSTANCES

| SUBSTANCE<br>(UNIT OF MEASURE)              | YEAR<br>SAMPLED | MCL<br>[MRDL] | MCLG<br>[MRDLG] | AMOUNT<br>DETECTED | RANGE<br>LOW-HIGH | VIOLATION | TYPICAL SOURCE                                                                                         |
|---------------------------------------------|-----------------|---------------|-----------------|--------------------|-------------------|-----------|--------------------------------------------------------------------------------------------------------|
| Arsenic (ppb)                               | 2013            | 10            | 0               | 0.67               | ND–2              | No        | Erosion of natural deposits; Runoff from orchards; Runoff from glass and electronics production wastes |
| Barium (ppm)                                | 2013            | 2             | 2               | 0.04               | ND–0.06           | No        | Discharge of drilling wastes; Discharge from metal refineries; Erosion of natural deposits             |
| Chlorine (ppm)                              | 2013            | [4]           | [4]             | 1.08               | 0.08–3.90         | No        | Water additive used to control microbes                                                                |
| Fluoride (ppm)                              | 2013            | 4             | 4               | 0.05               | ND–0.15           | No        | Erosion of natural deposits; Discharge from fertilizer and aluminum factories                          |
| Haloacetic Acids [HAAs]–Stage 1 (ppb)       | 2013            | 60            | NA              | 0.25               | ND–1              | No        | By-product of drinking water disinfection                                                              |
| Haloacetic Acids [HAAs]–Stage 2 (ppb)       | 2013            | 60            | NA              | ND                 | NA                | No        | By-product of drinking water disinfection                                                              |
| TTHMs [Total Trihalomethanes]–Stage 1 (ppb) | 2013            | 80            | NA              | 21.9               | ND–68.1           | No        | By-product of drinking water disinfection                                                              |
| TTHMs [Total Trihalomethanes]–Stage 2 (ppb) | 2013            | 80            | NA              | 1.17               | ND–5.6            | No        | By-product of drinking water disinfection                                                              |

Tap water samples were collected for lead and copper analyses from sample sites throughout the community

| SUBSTANCE<br>(UNIT OF MEASURE) | YEAR<br>SAMPLED | AL  | MCLG | AMOUNT<br>DETECTED<br>(90TH%TILE) | SITES ABOVE<br>AL/TOTAL<br>SITES | VIOLATION | TYPICAL SOURCE                                                       |
|--------------------------------|-----------------|-----|------|-----------------------------------|----------------------------------|-----------|----------------------------------------------------------------------|
| Copper (ppm)                   | 2013            | 1.3 | 1.3  | 0.790                             | 0/30                             | No        | Corrosion of household plumbing systems; Erosion of natural deposits |
| Lead (ppb)                     | 2013            | 15  | 0    | ND                                | 0/30                             | No        | Corrosion of household plumbing systems; Erosion of natural deposits |

### SECONDARY SUBSTANCES

| SUBSTANCE<br>(UNIT OF MEASURE) | YEAR<br>SAMPLED | SMCL | MCLG | AMOUNT<br>DETECTED | RANGE<br>LOW-HIGH | VIOLATION | TYPICAL SOURCE                                           |
|--------------------------------|-----------------|------|------|--------------------|-------------------|-----------|----------------------------------------------------------|
| Chloride (ppm)                 | 2013            | 250  | NA   | 30.3               | ND–61             | No        | Runoff/leaching from natural deposits                    |
| Iron (ppb)                     | 2013            | 300  | NA   | 270                | ND–400            | No        | Leaching from natural deposits; Industrial wastes        |
| Sulfate (ppm)                  | 2013            | 250  | NA   | 3.66               | ND–11             | No        | Runoff/leaching from natural deposits; Industrial wastes |

### UNREGULATED SUBSTANCES

| SUBSTANCE<br>(UNIT OF MEASURE) | YEAR<br>SAMPLED | AMOUNT<br>DETECTED | RANGE<br>LOW-HIGH | TYPICAL SOURCE              |
|--------------------------------|-----------------|--------------------|-------------------|-----------------------------|
| Hardness (ppm)                 | 2013            | 141                | 109–183           | Erosion of natural deposits |
| Sodium (ppm)                   | 2013            | 13                 | ND–25             | Erosion of natural deposits |

## Definitions

**AL (Action Level):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**MCL (Maximum Contaminant Level):** The highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology. Secondary MCLs (SMCLs) are established to regulate the aesthetics of water (i.e., taste and odor).

**MCLG (Maximum Contaminant Level Goal):** The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.

**MRDL (Maximum Residual Disinfectant Level):** The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

**MRDLG (Maximum Residual Disinfectant Level Goal):** The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**NA:** Not applicable

**ND (Not detected):** Indicates that the substance was not found by laboratory analysis.

**ppb (parts per billion):** One part substance per billion parts water (or micrograms per liter).

**ppm (parts per million):** One part substance per million parts water (or milligrams per liter).